

Nr	Symbol	1s	2s	2p	Elektron konfiguration	
1	H	↑			$1s^1$	$1s^1$
2	He	↑↓			$1s^2$	[He]
3	Li	↑↓	↑		$1s^2 2s^1$	[He] $2s^1$
4	Be	↑↓	↑↓		$1s^2 2s^2$	[He] $2s^2$
5	B	↑↓	↑↓	↑	$1s^2 2s^2 2p^1$	[He] $2s^2 2p^1$
6	C	↑↓	↑↓	↑ ↑	$1s^2 2s^2 2p^2$	[He] $2s^2 2p^2$
7	N	↑↓	↑↓	↑ ↑ ↑	$1s^2 2s^2 2p^3$	[He] $2s^2 2p^3$
8	O	↑↓	↑↓	↑↓ ↑ ↑	$1s^2 2s^2 2p^4$	[He] $2s^2 2p^4$
9	F	↑↓	↑↓	↑↓ ↑↓ ↑	$1s^2 2s^2 2p^5$	[He] $2s^2 2p^5$
10	Ne	↑↓	↑↓	↑↓ ↑↓ ↑↓	$1s^2 2s^2 2p^6$	[Ne]